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A6_1 Zap!.. and the Poles Are Gone!

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Abstract

We consider using the most powerful laser available to terraform Mars by sublimation of polar dry ice. The increase in atmospheric concentration of CO₂ results in a surface temperature increase of 2.7×10^{-5} K.

Introduction

August 8th 2016, also known as the "Overshoot Day", marked the point at which humanity began consuming more resources than the Earth could regenerate naturally [1]. With the growing population and the increasing energy demand, the idea of colonising space in the near future will become a necessity. Due to its close proximity and the many similarities it shares with our home planet, Mars stands out as the most obvious candidate to establish a colony on. However a major hurdle is the extremely cold surface temperature of 218K [2]. We researched the possibility of raising the surface temperature by sublimating the dry ice (CO₂) present at the Martian poles using the most powerful laser currently available.

Theory

Both polar caps are composed primarily of water ice with a thin layer of dry ice covering the top [3]. The mass of dry ice present at the north pole, $M_n = 1.3 \times 10^{15}$ kg, was calculated using the thickness of dry ice, $d_n = 1$ m, the radius, $r_n = 500$ km [4], and its density, $\rho = 1600$ kgm⁻³ [5] as:

$$M_n = \pi r_n^2 d_n \rho \quad (1)$$

Consequently the energy required to completely sublimate any dry ice, Q_n , can be calculated as:

$$Q_n = M_n c \Delta T + L_s M_n \quad (2)$$

where $c = 0.709$ kJkg⁻¹K⁻¹ [6] is the specific heat capacity of dry ice at 175K, $\Delta T = 74.5$ K [2] is the temperature difference between the minimum annual surface temperature at the poles (120.15 K), and the sublimation point of CO₂ (194.65 K) and $L_s = 571$ kJkg⁻¹ [5] is the latent heat of sublimation. This process is also used to get the energy required to sublimate the southern pole, Q_s , using $d_s = 8$ m and $r_s = 175$ km [4], giving a CO₂ mass of $M_s = 1.2 \times 10^{15}$ kg.

The sublimation of the poles' dry ice could lead to a significant increase in the concentration of CO₂ within the atmosphere, resulting in the increase of the surface temperature due to a stronger greenhouse effect.

This influx of CO₂ will increase its radiative forcing, ΔF , which is the difference between the energy absorbed and radiated by Mars [8]. The change in surface temperature, ΔT_{surf} , can be found via:

$$\Delta T_{surf} = \lambda \Delta F \quad (3)$$

where λ has a value of 0.8 Km²W⁻¹ [8]. ΔF (Wm

-2) can be found using:

$$\Delta F = 5.35 \ln\left(\frac{C}{C_0}\right), \quad (4)$$

where C_0 and C are the CO_2 concentrations in parts per million by volume (ppmv) in the atmosphere before and after sublimation respectively. The Martian atmosphere is composed of 95.97% CO_2 . The percentage volume of CO_2 in the atmosphere gives the initial concentration as $C_0 = 959700$ ppmv, while the final concentration is found via:

$$C = \frac{0.96V + V_a}{V + V_a} \quad (5)$$

where $V_a = M_T/\rho_{\text{CO}_2}$ is the volume of CO_2 added to the atmosphere, ρ_{CO_2} is the density of CO_2 (1.977 kg m^{-3}), $V = 9 \times 10^{18} \text{ m}^3$ [7] is the volume of the atmosphere and $M_T = M_n + M_s$ is the total mass sublimated from both poles.

Results

The final CO_2 concentration has a value of $C = 950006$ ppmv. The total energy required to sublimate both poles is $Q = Q_n + Q_s = 1.6 \times 10^{21} \text{ J}$ where $Q_n = 8.1 \times 10^{20} \text{ J}$ and $Q_s = 7.5 \times 10^{20} \text{ J}$. The LFEX laser is currently the most powerful laser in the world with a peak output power of 2PW, delivering 150-200J in one picosecond [9]. In optimal conditions such as complete absorption, a laser would be able to sublimate both CO_2 reservoirs in ~ 9 days with a continual output. Using Equation 3 we found that a CO_2 input of $1.3 \times 10^{15} \text{ kg}$ into the atmosphere would lead to a surface temperature increase of $2.7 \times 10^{-5} \text{ K}$.

Discussion

The energy required to sublimate the poles is beyond human capabilities; this value is an underestimate due to simplified assumptions. The temperature change calculated is unrealistic as we have ignored re-condensation of CO_2 and atmospheric stripping due to solar wind. Another consideration is the evaporation of water

ice which is 1000 times more abundant than dry ice. Evaporation of this would release a large amount water vapour into the atmosphere. Water vapour acts as a greenhouse gas and drives climate change much in the same way as CO_2 . The only difficulty in achieving such a feat is the length of time required, which is calculated to be ~ 1700 years.

Conclusion

The prospect of using the LFEX laser for terraforming Mars is not feasible. The power required, and the technological limitations of producing a continuously firing 2PW laser, prevents this technique from being practical.

References

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